Printed Pages: 1 Roll No. RPH204

B.PHARM.

THEORY EXAMINATION (SEM–II) 2016-17 PHARMACEUTICAL ANALYSIS-I

Time: 3 Hours Max. Marks: 70

Note: Be precise in your answer. In case of numerical problem assume data wherever not provided.

SECTION - A

1. Explain the following:

 $7 \times 2 = 14$

- (a) Accuracy
- **(b)** Secondary standard
- (c) Solubility product
- (**d**) pH
- (e) Law of mass action
- **(f)** Complexing agents
- (g) Titrant

SECTION - B

2. Attempt any five of the following questions:

 $5 \times 7 = 35$

- (a) Define buffer. Discuss mechanism of Acidic buffer.
- **(b)** Describe method for determination of hardness of water.
- (c) Discuss Mohrs method for Complexometric titrations.
- (d) Discuss Kjeldahls method of nitrogen determination.
- (e) Explain requirements for primary and secondary standards with suitable examples.
- **(f)** Discuss titrations involving Ceric sulphate.
- (g) Describe Handerson-Hasselbach equation.
- (h) Discuss various methods for expressing concentration.

SECTION - C

Attempt any two of the following questions:

 $2 \times 10.5 = 21$

- **3.** Discuss different techniques of analysis and explain significance of quantitative analysis in quality control.
- **4.** Discuss principle, requirements and methods for Iodometry and Iodimetry.
- 5. Write a note on Radio assays and discuss Assay of sodium iodide (¹³¹I) solution as per IP.