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PAPER ID : 5052	Roll No.				

B. PHARM.

(SEM. IV) EXAMINATION, 2006-07 PHARMACEUTICS - IV UNIT OPERATIONS - II

Time: 3 Hours] [Total Marks: 80

Note: (1) Answer all questions.

- (2) All questions carry equal marks.
- 1 Attempt any **four** of the following: 4×4
 - (a) Discuss the importance of unit operation in pharmaceutical technology.
 - (b) Write note on Primary and Secondary quantities
 - (c) The dimensional formula of a heat transfer co-efficient h is [h] = $QL^{-2} \theta^{-1} T^{-1}$.

In an experiment h was found to be 200 Btu/h (sq.ft)°F. What is the value in kcal/(sq.m) (°C) (hr)?

- (d) Discuss the following:
 - (i) Mole fraction
 - (ii) Graphical integration.
- (e) Write the application of dimensional analysis to heat transfer by natural and forced conviction equations.

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2 Answer any **four** of the following:

- 4×4
- (a) Discuss the factors to be considered in the selection of an evaporator.
- (b) Discuss the classification of pharmaceutical evaporators with their applications.
- (c) Discuss the advantage, disadvantage and application of steam jacketed kettle with diagram.
- (d) Write note on following:Steam traps and their use in evaporation.
- (e) Which equipment would you suggest for the continuous evaporation of large volume of an aqueous solution containing thermolabile constituents? Give reason.
- 3 Attempt any **two** of the following: 8×2=16
 - (a) Explain briefly boiling point diagram of the following mixtures:
 - (i) Alcohol water
 - (ii) Benzene Toluene.

Discuss the behaviours of these mixtures during distillation.

- (b) Discuss the construction and operational principle of large scale vacuum distillation unit. What are the advantage, difficulties and application in pharmacy.
- (c) Derive Rayleigh's equation and write an experimental procedure to verify it. Discuss the principle of rectification.

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4 Answer any **four** of the following:

- 4×4
- (a) Distinguish an atmosphere compartment dryer from vacuum compartment dryer.
- (b) Comment briefly on the following statements:
 - (i) Freeze dried materials have very good solubility
 - (ii) IR drying can be used for drying aqueous solutions and sticky masses.
- (c) Derive the equation of q-line

$$Y = [-q/(1-q)]. X + X_F/(1-q)$$

- (d) Comment on the following:
 - (i) Moisture content and L.O.D.
 - (ii) Spray drying yields a porous product
- (e) Compare freeze drying with spray drying.
- 5 Attempt any **two** of the following:
- 8×2
- (a) Discuss briefly the devices employed for measurement of temperature and pressure.
- (b) Write note on:
 - (i) Thermocouples
 - (ii) Pyrometers
 - (iii) Level measuring devices
 - (iv) Ionization gauges.
- (c) Distinguish the following
 - (i) Analog type Vs Digital type
 - (ii) Plug flow reactor *Vs* Continuous stirred tank reactor.

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